

Studies on fat in some chicken products with trial to prevent its oxidation

El- Shater, M.A*; Hanan, G. Seadawy** and Maha, M. Mohamed**

* Animal Health Research Institute (AHRI), Department of Food Hygiene

** Department of Chemistry Food Deficiency

Soshater 2020 @ yahoo. Com

Abstract

A total of 100 frozen half cooked chicken nuggets and shawerma (50 of each) were collected from grocery markets in Giza and Cairo governorates. The samples were subjected to chemical analysis (Fat %, Cholesterol, Acid value %, free fatty acids%, peroxide value m Eq/ kg and TBA value mg (MD/ Kg). The mean values of the aforementioned parameters for nuggets and shawerma were 17.78 ± 0.36 , 64.32 ± 0.94 , 1.88 ± 0.14 , 0.95 ± 0.07 , 10.13 ± 0.52 , 0.71 ± 0.052 and 15.3 ± 0.28 , 57.98 ± 0.69 , 1.85 ± 0.13 , 1.1 ± 0.17 , 10.912 ± 0.064 , 0.74 ± 0.035 respectively. There were significant decrease of fat % and cholesterol (mg/ 100 gm) at $P < 0.01$ in chicken shawerma lower than chicken nuggets. According to **ES (2005) and Codex Standard (1991)**, the rejected samples percent of both nuggets and shawerma were 84 and 86 which exceeded the recommends fat percent. On the other hand, the percent of rejected nuggets and shawerma due to the fat oxidation criteria exceeding limits (Acid value, free fatty acids, peroxide value and TBA) were 6 and 8 respectively. There is significant increase in shelf life at $P < 0.01$ in the treated chicken nuggets and shawerma with ascorbic acid more than untreated samples when stored at (-18°C) . The results were statistically analysed and it was recommended to use natural antioxidant for preventing the fat oxidation and extending the shelf life during freezing storage.

Key words: Chicken Products-Fat Oxidation-TBA value-Nuggets-Shawerma.

Introduction

Egypt's modern poultry industry began in 1964 with establishment of the national General poultry company (G.P.C) which aimed to provide Egypt's fast growing human population with high quality a fordable animal protein. Another important goal for G.P.C was the transformation of poultry production into an industry rather than an agricultural activity, through the introduction of modern technology of poultry processing and skilled management to get different chicken product as chicken shawerma, chicken nuggets etc. (**FAO, 2006**). Chicken nuggets are one of chicken products consists of chicken meat with seasoning with different shape (each piece $16 \pm 2\text{g}$), preferred (half cooked) with light golden brown and breaded. It must be free from banned antibiotics, pesticides, residues and sanitizer residues, the product is stored at -18°C (**Venky's, 2010**).

Shawerma is a well known precooked product in Middle East countries from beef or chicken

meat with animal fat the product kept in freezing state at (-18°C) fat (**Khalid, 2002**).

For poultry, pre-cooking also represents a "Pasteurization" process to improve the microbiological status—relative particularly to salmonella SPP. Many products are cooked to a center temperature above 70°C . The meat – products uses an enormous range of cooking methods but for poultry the use of fryer and ovens predominant (**Singhal et al., 2001**).

Freezing is as excellent process for preserving the quality of poultry and poultry products as well as other meat for long periods. However, deterioration of quality caused by chemical or physical factors can occur. Many studies have shown that lipid oxidation is one of the primary causes of quality losses of frozen stored poultry and meat products (**Gökalp et al. 1983**).

The most important process in the oxidation of lipids in meat and poultry products is the peroxidation of polyunsaturated fatty acids from cell membrane. The major catalysts are transi-

tional metal ions, such as Fe^{2+} and Cu^+ . Heme compounds of meat can also contribute to this process due to the participation of heme-iron in accelerating lipid peroxidation (**Decker and Welch 1990; Pikul 1992**).

During food processing and storage, polyunsaturated fatty acids tend to be oxidized. Cholesterol can be oxidized by the same mechanism as fatty acids. Therefore, lipid radicals formed during the processing and storage of foods can accelerate the oxidation of cholesterol and produce cholesterol oxidation products (COPs) (**Chan *et al.*, 1993**).

Aerobic packaging significantly increased cholesterol oxidation products (COPs) and thiobarbituric acid relative substances (TBARS) in cooked poultry and meat products after 7 days storage which were closely related to the fatty acid composition of food (**Du *et al.*, 2001**).

Acid value test measures free fatty acid as an indication of hydrolytic rancidity. Peroxide value is one of the most widely used tests for oxidative rancidity; peroxide value is a measure of the concentration of peroxides and hydroperoxides formed in the initial stages of lipid oxidation. The 2 – thiobarbituric acid (TBA) method is the most widely used test for measuring the extent of lipid oxidation in muscle foods. The test is believed to measure the breakdown products of unsaturated fatty acids oxidation. Saturated aldehydes, 2 – enals, and 2-dienals, produced in the termination phase of lipid oxidation, can be detected by reaction with 2 – thiobarbituric acid (**Haimd, 2011**).

Biologically active compounds can be destroyed and, in some cases, toxic and carcinogenic substances represented by hydroperoxides, acids, etc. accumulate (**Balev *et al.*, 2005**).

The meats are important sources of fat in the typical diet in the world, many consumers believe that red meat is unhealthful, because it is high in saturated fatty acids (SFA) and cholesterol. Recently, it has been demonstrated that replacement of red meat with chicken is associated with significant decrease in total cholesterol levels in

diabetic patients, this effect is probably related to the higher PUFA (polyunsaturated fatty acids) content of chicken meat in comparison to

beef (**Gross *et al.*, 2002**).

diabetic patients, this effect is probably related to the higher PUFA (polyunsaturated fatty acids) content of chicken meat in comparison to beef (**Gross *et al.*, 2002**).

This study was planned to study:-

Part I: Chemical criteria of market half cooked frozen chicken nuggets and shawarma including.

1- Fat %.

2- Cholesterol (mg / 100 gm).

3- Fat Lipolysis.

3-1- Free fatty acid % as oleic.

3-2- Acid value % as oleic.

4- Fat oxidation criteria.

4-1- Peroxide value mEq/ kg.

4-2- Thiobarbituric acid value (TBA mg MD/kg).

Part II

Trial to prevent the lipolysis and fat oxidation through addition of Ascorbic acid as antioxidant (natural) to experimentally produced chicken nuggets and shawarma, then stored at -18°C with control samples (without Ascorbic acid).

The experimentally produced chicken nuggets and shawarma (treated with ascorbic acid) and control chicken nuggets and shawarma (untreated with ascorbic acid) were examined every 2 weeks till spoilage for free fatty acids, acid value, peroxide value and TBA value.

Materials and Methods

Part I:-

A total of 100 frozen chicken nuggets and shawarma (50 of each) were collected from markets of Giza and Cairo governorates. The samples were transported to the laboratory in ice box without delay for the following examinations.

1. Determination of fat %.

The test was carried out according to (**AOAC, 2000**), in which the traditional technique is the quantitative determination of fat through a solvent extraction with petroleum ether. The soluble material is extracted from dried tested samples by a two – steps treatment with petroleum ether solvent (Soxhlet extraction procedure).

2. Determination of cholesterol content:

Cholesterol was determined according to **Ojia-ko and Akubugwo (1997)** using spectrophotometer. Only 0.1 ml of extracted fat and standards cholesterol dissolved in chloroform in ratio 1: 10 was evaporated to dryness in water bath at 50 °C. Glacial acetic acid (3 ml) and 3.0 ml color reagent was added to each sample and standards, then shaking vigorously to dissolve the sample. Blank contained 2.0 ml chloroform, 3 ml Glacial acetic acid and 3 ml of color reagent was prepared.

After cooling for 30 min. at room temperature, absorbance of blank sample and standards were measured at 560 nm. The cholesterol content in sample were calculated from the standard curve.

3. Determination of Lipolysis criteria as follows:

3.1. Acid Value:

It was carried out according **IUPAC (1979)** using titration technique against alkali (1 – 10 gm dissolved fat). The acid value was calculated according to the equation.

$$\text{Acid value (\%as Olic)} = \frac{\text{Titration (ml) } 5.61}{\text{Wt of sample used}}$$

3.2. Free Fatty acids:

The official titration with alkali method recommended by **ISO (1980)** (Reference method) was used the F.F.A figure in usually calculated as oleic (percent).

4. Determination of Fat oxidation criteria:

4. 1 Peroxide Value:

The sample was extracted for fat. To quantify peroxide value, (0.01– 0.05g) was placed in 10 ml screw capped test tube and dissolved in: 1 ml chloroform / acetic acid (2: 3), with addition of 100 ml Fe (II) solution, mix for 15 sec. on vortex mixer and left it in dark place for 10 min. Deionized water (2 ml) was added and 4 ml of diethyl ether (containing 7 ppm BHT). One ml of aqueous phase was mixed with 100 ml of saturated ammonium thiocyanate solution. The sample was measured against a water

blank at wave length 470 nm. after 10 min.

The peroxide value was measured using calibration curve. For calibration, a set of Fe (III) concentration in range 0 – 10 µg / ml. The method was carried out according to **Shantha and Decker (1994)** the P.V was expressed as mEq / kg.

4. 2 Thiobarbituric acid value:

The method according to **Tarlidgis et al (1960)** was used, in which the reaction is applied to a distillate produced under standardized conditions from an acidified macerated food. The results are expressed as malonaldehyde by reference to a standard graph prepared by using 1, 1, 3, 3 – tetra ethoxy propane which yields malonaldehyde by acid hydrolysis TBA no (as mg malonaldehyde/ Kg) = 7.8 D where D (absorbance against blank at 358 nm).

Part II:

1. Production of chicken Nuggets experimentally:

It was carried out according to **Yavas and Bilgin (2010)** using fresh poultry breast meat in addition to seasoning (control), the other group was treated with antioxidant (Ascorbic acid 500 ppm).

The samples (control and treated) were fried in vegetable oil for 2 – 3 min (half cooked). The control and treated chicken nuggets were put in suitable polyethylene bags then put in carton package.

2. Production of chicken shawerma experimentally:

It was carried out according to **Odu and Akamo (2012)**. Fresh boneless chicken breast was sliced, put in a pan, seasoned with salt, curry, thyme, black paper, nutmeg and garlic (control). The treated sample was made as mentioned in addition to (500 ppm) of ascorbic acid was added. The control and treated prepared shawerma were heated for 10 min (half cooking) after addition of water and vegetable oil. After cooling, control & treated shawerma were wrapped as mentioned in chicken nuggets.

3. Storage of Experimental samples (chicken Nuggets and chicken shawerma):

The Experimental samples were kept at -18°C and examined every 2 weeks for lipolysis and

fat oxidation criteria till spoilage of the products appear as mentioned before.

4-Statistical Analysis: Data obtained were statistically analyzed using **SPSS14 (2006)**.

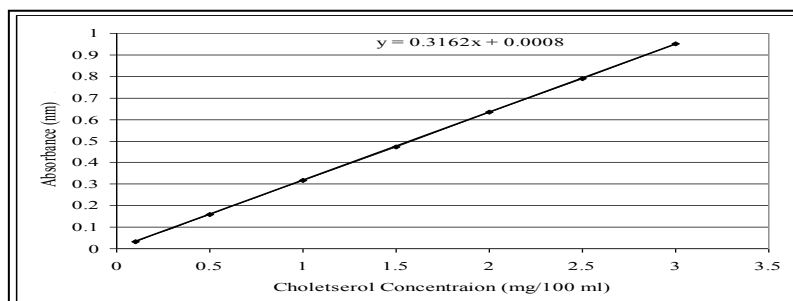


Fig. (1). Cholesterol standard curve.

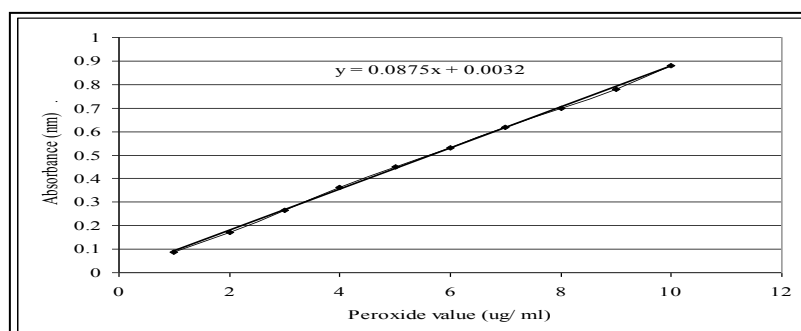


Fig. (2). Peroxide standard curve .

Results

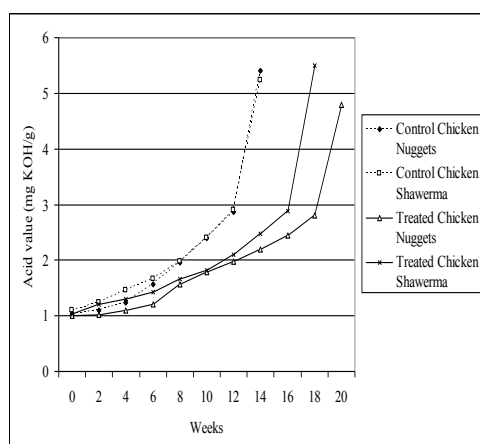
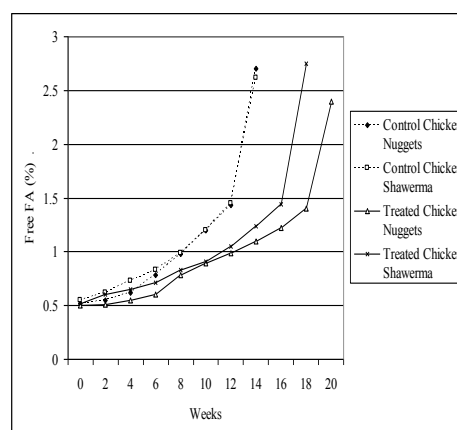
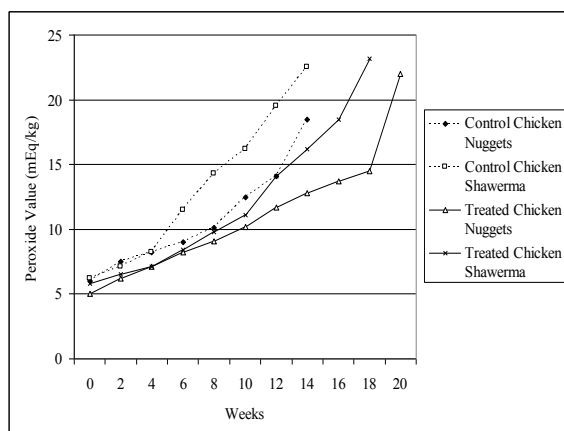
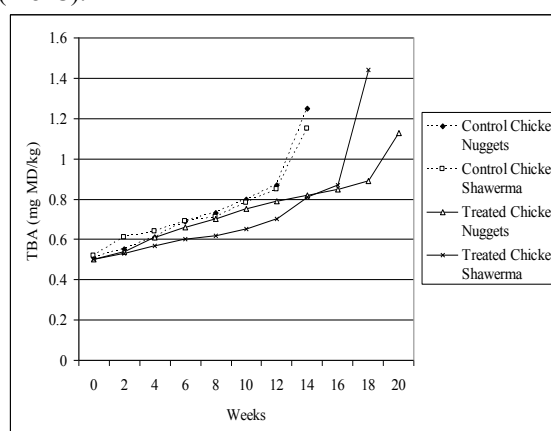
Table (1). Fat and cholesterol percent of market frozen chicken nuggets and shawerma.

		Chicken Nuggets		Chicken Shawerma	
		Fat (g/100 gm)	Cholesterol (mg/100 gm)	Fat (g/100 gm)	Cholesterol (mg/100 gm)
Rejected	No	42	0	43	0
	%	84	0	86	0
Accepted	No	8	50	7	50
	%	16	100	14	100
Min		14	55	12	50
Max		24	79	19	70
Mean		17.78	64.32	15.3*	57.98*
SE		0.363	0.936	0.282	0.693

*Significant at $P < 0.01$ using t-student test comparing with Chicken Nuggets

Table (2). Statistical analytical results of chemical analysis (lipolysis and fat oxidation) of market frozen chicken nuggets and shawerma (n = 50 of each).

			Acid value	Free FA	Peroxide Value	TBA
Chicken Nuggets	Rejected	No	3	3	3	3
		%	6	6	6	6
	Accepted	No	47	47	47	47
		%	94	94	94	94
	Min		1.02	0.56	6.2	0.52
	Max		5.9	2.95	23	1.44
	Mean		1.8832	0.9512	10.13	0.7104
	SE		0.14	0.07	0.52	0.025
Chicken Shawerma	Rejected	No	3	5	4	4
		%	6	10	8	8
	Accepted	No	47	45	46	46
		%	94	90	92	92
	Min		1.1	0.55	6	0.5
	Max		6	9	28	1.9
	Mean		1.8504	1.1	10.912	0.7404
	SE		0.13	0.17	0.064	0.035

**Fig. (3).** Changes of acid value of chicken nuggets and shawerma during freezing storage at (-18°C).**Fig. (4).** Changes of free fatty acids of chicken nuggets and shawerma during freezing storage at (-18°C).**Fig. (5).** Changes in peroxide value of chicken frozen nuggets and shawerma during freezing storage at (-18°C).**Fig. (6).** Changes in TBA value of chicken frozen nuggets and shawerma during freezing storage at (-18°C).

Discussion

In present study, the market samples of nuggets had mean values of fat % and cholesterol (mg / 100 gm) were 17.78 ± 363 & 64.32 ± 0.936 while the mean values for chicken shawerma (frozen) were 15.3 ± 0.282 and 57.98 ± 0.693 respectively the fat percent and cholesterol content (mg/ 100 gm) of frozen chicken shawerma were significantly lower than frozen chicken nuggets at $P < 0.01$.

Concerning the fat %, it was found that 8 samples of chicken nuggets (16%) and 7 samples of chicken shawerma (14%) were only accepted according to maximum fat (15% for nuggets and 12% for shawerma) recommended by **ES (2005)** table (1).

Microbial activities are considered the major factor of foods alteration during storage and manipulation, when these activities are under efficient control, the foods alteration will be of chemical nature, lipid oxidation is a principal chemical change of foods, which depends on the level of oxygen, (energy/ heat) and metals. Lipid oxidation products are responsible for the development of rancidity by the production of low molecular weight compounds that cause undesirable flavor (**Frankel, 1985; Frankel et al., 1987**), thus affecting the quality and limiting shelf life of food products. Cholesterol is also oxidized in similar reaction mechanisms to those observed of fatty acids. Many of the cholesterol oxidation products have adverse effects such as cytotoxicity and modifications of enzyme activity (**Erickson et al., 1978; Sevanin and Petreson, 1986; Bosinger et al., 1993**) atherosclerosis (**Kumar and Singal, 1991**) carcinogenicity and mutagenicity (**Ansari and Smith, 1979**).

In this study there was positive correlation between the fat content in both chicken nuggets and shawerma and their cholesterol content. This result was in accordance with that reported by **Dinh et al (2008)**.

Major sources of cholesterol in the human diet are meat from domestic livestock, although seafood is also rich in cholesterol content in this study in both nuggets and shawerma of chicken ranged from (55-79) and (50-70) mg /

100 g. Nearly similar results were reported by (**Chizolini et al. 1999; Mourot and Hermier 2001; Pironen et al., 2002; Valsta et al., 2005; Bragagnolo, 2009** and **Honikel, 2009**) who recorded that the cholesterol content in poultry products ranges from 40 – 90 mg / 100 gm. Higher concentration of cholesterol content was recorded by (**USDA, 2011**) for cooked chicken dark meat.

Table (2) illustrated the acid value, free fatty acids, peroxide value and TBA for both chicken nuggets and chicken shawerma. The chicken nuggets mean values of the mentioned parameters were 1.88 ± 0.14 , 0.95 ± 0.07 , 10.13 ± 0.52 and 0.71 ± 0.025 while for chicken shawerma were 1.8504 ± 0.13 , 1.1 ± 0.17 , 10.912 ± 0.064 and 0.74 ± 0.035 respectively.

According to the maximum TBA (mg MD/ kg) recommended by **ES (2005)** (0.9 mg MD/ kg). It was found that 3 samples (6%) of chicken nuggets and 4 samples (8%) of chicken shawerma were rejected due to rancidity.

The acid value and the free fatty acid of control and treated chicken nuggets and shawerma experimentally produced and stored(-18) were shown in Figs (3 & 4).

The treated chicken nuggets and chicken shawerma exceeded the acid value recommended by **codex standard(1991)** limit at the weeks 20, 18 weeks respectively.

It was noticed that the control chicken nuggets and shawerma were within limit (0.5 – 1.2) percent as oleic recommended by **IUPAC (1979)** till the week (12) of frozen storage while treated nuggets and shawerma till the week 18 and 16 respectively.

Free fatty acids are the products of enzymatic or microbial degradation of lipids and determination of FFA give information about the stability of fat during storage, this agrees with that reported by (**Das et al., 2008**). The free fatty acids of antioxidant treated nuggets and shawerma were significantly lower than control (non-treated) throughout the storage period, this result agreed with that reported by (**Yavas and Bilgin 2010**).

The experimentally produced chicken nuggets and chicken shawerma (with antioxidant) were

stored with experimentally control (without antioxidant) at -18°C till spoilage criteria of lipolysis and fat oxidation appear. The acid value of chicken nuggets and chicken shawerma of both experimentally control and treated at zero day were 1.04, 1, 1.1, 1.04 respectively. The acid value of chicken nuggets and chicken shawerma of control samples exceeded the maximum limit recommended by **Codex (1999)** (max 4.0 mg KOH/ g fat) at week (14). The initial peroxide value of the control and treated chicken nuggets and shawerma stored at -18°C at zero week were 6, 5 and 6.2 and 5.8 respectively. The peroxide value gradually increased throughout the storage period. At weeks 2, 4, 6, 8, 10, 12 and 14, the peroxide values of chicks nuggets control and treated were 7.5, 8.2, 9.5, 10.1, 12.5, 14.1, 18.5 and 6.2, 7.1, 8.2, 9.1, 10.2, 11.7, 12.8 respectively. It was noticed that at 4th week the peroxide value of the control nuggets exceeded the maximum limit recommended by **Codex Standard (1999)** (max. 12.5 m Eq of active O₂/ kg). The treated nuggets exceeded this limit at 20th week.

On the other hand, the peroxide value of chicken shawerma exceeded the maximum limit for control at the 10th week while at 16th for treated shawerma (Fig., 5). The peroxide value is the most commonly used parameter to measure lipid hydroperoxides. It also named primary lipid oxidation products. Chicken products have considerably fat and particularly vulnerable to lipid oxidation which leads to quality and nutritional value deterioration (**Olafsdottir et al., 1997; Yimaz, 1998**). The peroxide values were significantly increased at $P < 0.05$ during storage period in both nuggets and shawerma, this result in current study is in accordance with that of **Sal-lam (2007) and Yavas and Bilgin (2010)**. Nutritionist and buyers had established maximum peroxide value levels between 5 – 20 meq. O₂ / kg of fat (**Hamilton and Kirestein, 2008**).

The TBA (mg MD/ kg) of control nuggets and shawerma at weeks 0, 2, 4, 6, 8, 10, 12, 14, were 0.5, 0.52, 0.61, 0.69, 0.73, 0.8, 0.87, 1.21 and 0.52, 0.61, 0.64, 0.69, 0.71, 0.78, 0.85, 1.15 respectively. The TBA exceeded the maximum

limit (0.9 mg MD/ kg) recommended by **ES (2005)** at 14th week in both control chicken nuggets and shawerma.

The use of antioxidant (Ascorbic acid) extend the shelf life of treated chicken nuggets and shawerma to 18th, 16th weeks at which the TBA were within the permissible limit recommended by **ES (2005)**. At the weeks 20 and 18 the treated chicken nuggets and shawerma had TBA 1.13 and 1.44 mg MD/ kg which exceeded the maximum limit of TBA respectively (Fig., 6). The treatment of chicken nuggets and shawerma had statistically significant effect ($P < 0.05$) on TBA value during storage at - 18 C°. The lowest TBA in treated chickens nuggets and shawerma with ascorbic were 0.5 mg/ MD/ kg for each, while the highest TBA values were 1.13 and 1.44 mg MD/ kg. TBA values were significantly lower ($P < 0.05$) in all treated samples of nuggets and shawerma as compared to the control. This results agreed with that reported by **Ozgul and Cemalettin (2010)**.

In industrial processing, mainly synthetic antioxidant such as butylated hydroxytoluene, butylated hydroxyinsol and propylated gallate used prolongue the storage stability of food. However the demand of natural antioxidants has recently increased because of the toxicity and the carcinogenicity of synthetic antioxidants (**Juntachote et al 2006**). In addition, synthetic antioxidants have limited application because of their low water solubility (**Bekhit et al, 2003**). BHA, however, has a high solubility in animal fats (**Allen & Hamilton 1994; Copen 1994**). In meat processing, the Ascorbic acid is used as antioxidant (natural antioxidant) for preventing the off – flavor (**Bauernfiend 1985**) as well as maintaining the cured color.

The anti-oxidation mechanism of ascorbic acid is to activate the primary antioxidants, to be an oxygen scavenger, and to inactivate antioxidants (**Bauernfiend and Pinkert, 1970**). At low concentrations, ascorbic acid promotes lipid oxidation; however, at high concentrations it inhibits lipid oxidation (**Decker and XU, 1998**).

References

- Allen J. C. and Hamilton R. J. (1994).** "Rancidity in foods" 3rd ed. Blackie Academic and professional Chapman & Hall, London: 318 – 319.
- Ansari, G.A.S. and smith, L.L (1979).** "High performance liquid chromatography of cholesterol antioxidation products." J. chromatography. 175, 307 – 315.
- AOAC official Methods (2000)** Fat (crud) in meat and poultry products solvent extraction method AOAC International 17 ed Gaithersburg, MD chapter 39, 985. 15.
- Balev, D., Vulkova T., Dragoev S., Zlatanov M. and Bahtchevanska S. (2005).** "A comparative study on the effect of some antioxidants on the the lipid and pigment oxidation in dry fermented sausages." International Journal of Food Science Technology, 40: 977 – 983.
- Bauernfiend J.C. (1985).** "Antioxidant functions of L-ascorbic acid in food technology." International for vitamin and Nutrition Research (supplement) 27: 307 – 333.
- Bekhit A.E.D., Geesink G.H., Ilian M.A., Morton J. D. and Bickerstaffe R. (2003).** "The effects of natural antioxidants on oxidative processes and metmyoglobin reducing activity in beef patties." Food chemistry 81: 175 – 187.
- Bosinger; S., Luf, W. and Brandi, L. (1993).** "Oxysterols: Their occurrence and biological effect." Int. Dairy J. 3. 1 – 33.
- Bragagnolo N. (2009).** "Cholesterol and cholesterol oxides in meat and meat products." In: Nolle LML. Toldra F, editors. Handbook of muscle foods analysis Florida. CRC Pres. P 187-2012.
- Chan, S.H.; Gray J.I.; Gomaa E.A., Harte B.R., Kelly P.M. and Buckely, D.J. (1993).** "Cholesterol oxidation in whole milk powders as impudence by processing and packaging." Food Chem., 47: 321 – 328.
- Chizolini R. Zanardi E, Dorigoni V. and Ghidini S. (1999).** "Calorific value and cholesterol content of normal and low – fat meat and meat products." Trends Foods Sci 10: 119 – 28.
- Codex (1999).** "Codex standard for edible fats and oils codex" stan 19 – 1981, Rev. – 1999.
- Coppen P.P (1994).** "The use of antioxidants" In: Allen J.C., Hamilton R.J (eds). Rancidity in Foods: Blackie Academic and professional. Chapman & Hall, London: 318 – 319.
- Das, A.K., A.S.R. Anjaneyulu, Y.P.Gadexar, R. P. Singh and H. Pragati, (2008).** "Effect of full – fat soy paste and textured soy granules an quality and shelf – life of goat meat nuggets in frozen storage." Meat Sc., 80.
- Decker E. A. and Welch, B. (1990).** "Role of Ferrition as lipid oxidation catalyst in muscle food." J. of Agricultural and food chemistry, 38: 674 – 677.
- Decker E. A. and Xu, Z. (1988).** "Minimizing rancidity in muscle foods." Food Technology, 52 (10). 54-59.
- Dinh, T.T.N.; Blanton JR; Jr, Brooks JC.Miller MF. and Thompson, L.D. (2008).** "A simplified method for cholesterol determination in meat and meat products." J food comp Anal. 21 (4) 306 -14.
- Du, M., Nam, K.C., and AHN, D. U. (2001).** "Cholesterol and lipid oxidation products in cooked Meat as affected by raw – meat packaging and irradiation and by cooked – meat packaging and storage time." Journal of food science vol. 66, No. 9.
- Erickson, S.K., Matsui, S.M., strewsbury, M.A., cooper, A.D., Gordan, R. (1978).** "Effect of 25 – hydroxylcholesterol on rat hepatic.
- ES (2005).** "Egyptian standard products of meat poultry treated with heat." No. 3493.
- FAO (2006).** "Poultry sector country." Egypt FAO Animal production and Health Division Emergency center for transboundary Animal Diseases Socio Economics, production and Biodiversity unit.
- Frankel, E.N, Nash A. M., Snyder, J.M (1987).** "A methodology study to evaluate quality of soybeans stored at different moisture levels." J. Am. Oil chem. Soc. 74, 387 – 391.
- Frankel, E.N. (1985).** "Chemistry of free radical and singlet oxidation of lipids." Prog. Lipid Res., 23, 197 – 221.
- Gökalp, H.Y.; Ockerman H.W., Plimpton**

- R.F. and Harper, W.J. (1983).** "Fatty acids of neutral and phospholipids, rancidity scores and TBA values as influenced by packaging and storage." *Journal of food science.*, 48: 829 – 834.
- Gross, J.L; Zelmanvutz. Moulin, C.C., DF Mello, V; perassolo, M.; Leitaó.; Hoefel A., Paggi, A. and Azevedo, M.H., (2002).** "Effect of a chicken – based diet on renal function and lipid profile in patients with type 2 diabetes." *D. care*, V. 25, P 645 – 651.
- Haimd Raze Gheisari (2011).** "Correlation between acid, TBA, peroxide and iodine values, catalase and glutathione peroxidase activities of chicken, cattle and camel meat during refrigerated storage." *Veterinary world Research*, Vol. (4). 153 – 157.
- Hamilition, C. R and D. Kirstein, (2008).** "Does Rancidity as measured by peroxide value, affected animal performance?." Internet document, URL HH P//: www. Derlingii. Com / pdf.
- Honikel, K.O. (2009).** "Composition and calories." In: Nollé LML, Toldra F, editors. *Handbook of processed meats and poultry analysis*. Florida: CRC Press: P 195 – 213.
- ISO 1980 No. 1740-** Standard, Determination of the acid value of the fat, in International standards Animal and vegetable oils and fats. International organization of standardization. Geneva.
- IUPAC (1979)** Standard Methods for the analysis of oils, fat and derivatives 6th ed. Oxford: pogram on press.
- Juntachote T., Berghofer E., Sie Benhandl, S. and Bauer F. (2006).** "The oxidative properties of Holy basil and Galangal in cooked ground pork." *Meat science*, 72: 446 – 456.
- Khalid, A.L.I. (2002).** "Effect of two methods of grilling on the oxidative rancidity and cholesterol oxidation in beef and chicken shawerma." *Grasses Y Aceites* Vol. 53 Fasc. 3, 335 – 339.
- Kumar, N. and Singnal, O.P (1991).** "Cholesterol oxides and atherosclerosis: A review." *J. Sci Food Agric.*, 55, 497 – 510.
- Mourot, J. and Hermier D. (2001).** "Lipid in monogastric animal meat." *Reprod Nutr Dev* 41: 109 – 118.
- Odu, N.N., and Akamo, U.M. (2012)** the microbiological assessment. Of ready to eat food (shawerma) in port Harcourt city. *Nigeria Nature and science* 10 (18).
- Ojiako, O. A. and Akubugwo, E.I. (1994).** "An introductory Approach to practical Biochemistry ."CRC Publication, Owerri, PP: 132.
- Olafsdottir, G., E., Martinsdottir, J. Oehlen-schlager, P. Dalgaard, B. Jensen, I. Undeland, I.M. Mackie, G. Henehan, J. Nielsen and H. Nilsen, (1997).** "Methods to evaluate fish freshness in research and industry." *Trends in Food Sci. Tech.*, 8: 258 – 265.
- Ozgul Ozer and Cemalettin Saricoban (2010).** "The effect of Butylated Hydroxyinsol, Ascorbic Acid, and α -Tocopherol on sole quality characteristics of Mechanically Deboned chicken patty during freeze storage." *Czech J. food Sci.* Vol. 28: No 2, 150-160.
- Pikul J. (1992).** "The oxidation of lipids and the development of warmed over flavour in heated and stored meat, Part I." *Gospodarka Mie*, 07: 20 – 23 (In Polish).
- Pironen, V.; Toivo J. and Lampi A.M. (2002).** "New data for cholesterol contents in meat, fish, milk, eggs and their products consumed in Finland." *J. Food comp Anal* 15: 705 – 13.
- Sallam. K.I (2007).** "Antimicrobial and antioxidant effects on sodium acetate? sodium lactate and sodium citrate in refrigerated sliced salmon." *Food contr.*? 18: 566 – 575.
- Sevanian, A.and Peterson, A. R (1986).** The cytotoxic and mutagenic properties of cholesterol oxidation products. *Food Chem.. Toxicol.* 24, 1103 – 1110.
- Shantha, C., and E.A Decker (1994)** Rapid sensitive, Iron based spectrophotometric methods for Determination of peroxide value of food lipids *J. AOAC Int.* 7: 421 – 424.
- Singhal, R.S; Kulkarni, P. R and Rage, D.V (2001).** "Hand book of Indices of food quality and Authenticity." Food head publishing.
- SPSS 14 (2006).** "Statistical Package for So-

- cial Science, SPSS for windows Release 14.0.0, 12 June, 2006." Standard Version, Copyright SPSS Inc., 1989-2006, All Rights Reserved, Copyright © SPSS Inc.
- Tarlidgis, B.S.; Watts, B. M.; Younathan, M.T. and Dugan, L. (1960).** "Determination of thiobarbituric acid value in food." J. of American Oil Chemists Society, 37: 44.
- USDA (U. S. Department of Agriculture), Agricultural Research Service. (2011).** "USDA national nutrient database for standard reference." Beltsville, Med: USDALARS Nutrient Data Laboratory.
- Valsta, L.M.; Tapanainen, H. and Mannisto (2005).** "Meat fats in nutrition." Meat Sci. 70: 525 – 30.
- Venky`s (2010).** "Processed chicken nuggets." Venky`s publish – India.
- Yavas, E. and Bilgin, B. (2010).** "Effect of calcium Lactate, sodium diacetate and sodium chloride mixture on the micro biological, chemical and sensory properties of chicken nuggets stored in refrigeration and under modified atmosphere." Inter national Journal of poultry science 9 (1). 66 – 71, 2010.
- Yilmaz, I. (1998).** "Fark ambalajlama yontemi vedepolama sicakliginin tekirdag koftesinin bazimik- robiyolojik, fiziksel vekimyasal ozellikleriuzerine etkilerinin belirlenmesi." Doktora tezi. T. Uni. Zir. Fak. Tekirdag, PP: 204.

دراسات على الدهون في بعض منتجات الدواجن مع محاولة لمنع اكسدها

* محمد أحمد الشاطر - * حنان جودة سعداوى - ** مها محمود محمد

* معهد بحوث صحة الحيوان – قسم صحة الأغذية ** قسم الكيمياء والنقص الغذائي

المخلص العربي

تم فحص 100 عينة من كل من ناجيتس الدجاج وشاورما الدجاج المجمدة والموجودة بأسواق محافظة القاهرة والجيزة (50 عينة من كل نوع) كيميائياً (نسبة الدهن – نسبة الكوليستيرول – الأحماض الدهنية الحرة – قية الحمض – قيمة البيروكسيد – قيمة حمض الثابوباربتيوريك). كان متوسط قيم الاختبار السابقة لكل من الناجيتس والشاورما 17.78 ± 0.36 ، 63.32 ± 0.9 ، 0.14 ± 0.17 ، 1.88 ± 0.07 ، 0.95 ± 0.07 ، 10.13 ± 0.52 ، 0.71 ± 0.052 ، 15.3 ± 0.289 ، 57.98 ± 0.69 ، 1.85 ± 0.13 ، 0.74 ± 0.13 ، 10.912 ± 0.064 ، 1.099 على التوالي.

كان هناك نقص معنوي في نسبة محتوى الدهن في عينات شاورما الدجاج مقارنة بالناجيتس عند $P < 0.01$ وطبقاً لمحتوى الدهن وجد أن 43% من عينات الشاورما و40% من عينات الناجيتس قد كانت مرفوضة نظراً لاحتوائها على نسبة أعلى من الحدود التي قررتها المواصفات القياسية المصرية (2005). وقد وجد أن 6%، 8% من عينات الناجيتس والشاورما الموجود بالأسواق كانت مرفوضة لأنها تخطت الحدود المسموح بها لاجتيازات تخزين الدهون طبقاً للمواصفة المعدية (2005) ومواصفات الكودكس 1991.

وجد في الجزء الثاني من الدراسة أن إضافة حمض الاسكوريك للعينات التجريبية للناجيتس والشاورما قد زاد من فترة حفظها عند درجة حرارة التجمد (-18°C) معنوياً عند مقارنة العينات التجريبية التي لم يضاف إليها حمض الاسكوريك (عينات ضابطة).

وقد نوقشت النتائج إحصائياً وأوصت الدراسة بإضافة حمض الاسكوريك عند التصنيع لمنع اكسدة الدهون وزيادة فترة الصلاحية.